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$\text{CH}_3\text{CH}_2\text{COOH}(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{CH}_3\text{CH}_2\text{COO}^-(\text{aq}) + \text{H}_3\text{O}^+(\text{aq})$ Propanoic acid, $\text{CH}_3\text{CH}_2\text{COOH}$, is a carboxylic

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acid that reacts with water according to the equation above. At 25 C the pH of a 50.0 mL sample of 0.20 M CH

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