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Examiner(s): Dr Sarper Sarp Temperature: T in K = T in ° C +

273 Universal Gas Constant: R = 8.314 kJ kmol⁻¹ K⁻¹ Pressure: 1

MPa = 1 × 10⁶ Pa: 1 bar = 1 × 10⁵ Pa: 1 MPa = 9.869 atm: 1

atm = 760 mmHg Volume: 1 m³ = 1000 litres Power: 1 kW = 1

× 10³ W = 1 × 10³ J/s Conversion Factor: 1 kJ/kg = 1000 m² /s

2 Data Sheets are provided (see pages 7 - 10)

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