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Chapter 6 Standardized Test Practice 1. The actual number of atoms of each element contained in a molecule is given by the molecule's: A. empirical formula. C. mass percent formula. B. molecular formula. D. percent composition. 2. What is the molar mass of the sodium sulfate Na_2SO_4 ? F. 71.06 g/mol G. 94.05 g/mol H. 142.05 g/mol J. 215.13 g/mol 3.

Chapter 6 Standardized Test Practice

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Solubility of Various Solids Versus Temperature KNO₃ NaNO₃ NH₄Cl NaCl Na₂SO₄ KNO₃ NaNO₃ NH₄Cl Na₂SO₄ NaCl Key

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